

Properties of materials

Summer 2 2022 **Class Lynher**

To start the lesson we recalled what we already knew about properties of materials and wrote definitions for some of the key vocabulary. Most of the materials looked quite thin but it was surprising that they didn't let water go through them.

follow these instructions to test the pro

the surfac 1 to 5, will scratch with hardest to s Flexibility

> Then we used various methods to compare materials looking at properties such as magnetic, transparent and hardness.



For this lesson we focussed on the property of thermal insulation and did an investigation based on this.





It was weird how different materials worked differently to what I thought.



First of all we discussed the effect of resistance and did a practical demonstration to aid our understanding.

I was interested in how the light would shine at a different brightness depending on which material you put between the clips.

Then we designed an experiment to test how well different metals conducted electricity. We used a light sensor to help us get accurate results.





10.06 can investigate electrical onductors 150 e asonon it pin 245 paperclip 170 ken 160 200 200 old ring 160 180 silver rung 180 Housepoints hook



Then we designed and carried out an investigation about a factor in dissolving. I found it interesting that when we changed the temperature at the hot end there wasn't a big difference but at the cold end there was. First of all we discussed the vocabulary of soluble, insoluble and dissolve.

We found out that the hotter the water was the quicker something dissolved.

Dissolving Investigation

We discussed the vocabulary for different mixtures – linking to the previous lesson on dissolving.



It was really interesting that when we used the filter paper the water went really clear. Then we used various processes to separate materials from each other including filtration, sieving and evaporation.



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Irreversible changes

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A change is called irreversible if it cannot be changed back again.

For our final session we learnt about irreversible changes and then identified them in real life explaining how we knew it was irreversible.



We found out that if you cooked an egg it would go hard but when it is raw it is liquid.

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the change bate



funce wood is aun't it ethur turns to smoke or ash.



the code wintere changes by heating up and turning into a solid

what things had reverte changes

Science FLE	75/6 Proper	ges of materials	
Cultural capital Children see the real life value of science in life. Children are exposed to scientific vocabulary. Real-life jobs it could link to: electrical engineer, electrician, waste management.	TIMITE	Copper_ (Electrical Conductor) Plastic (Electrical Insulator)	Skills Plan different types of enquiry to answer questions Use scientific evidence to answer questions and support findings Use test results to make predictions Present findings from enquiries Record findings in a graph
Forever Facts	Exciting Books	Subject Sp	ecific Vocabulary
Most metals are both thermal and electrical conductors	Properties	materials	the substance something is made out of e.g.wood, plastic, metal
Wood and plastic are both thermal	Changes of Materials	conductor	a material that heat or electricity can easily pass through
		insulator	a material that does not let heat or electricity travel through
In an irreversible change you cannot get back to the original metorials		transparent	lets light through so the object can be looked through
Different materials are used for		soluble	will dissolve
different jobs based on their properties	sterce in integraptics	insoluble	will not dissolve
Sieving separates small particles		evaporating	liquid turns to gas or vapour
Filtering catches solid particles and	Our Endpoint	condensing	gas cools and turns to liquid
lets liquid through	I can describe the properties of materials using scientific vocabulary	melting	solid heats until it becomes liquid
What I have learnt before: Magnetism, electricity, states of matter		freezing	liquid cools and turns into solid